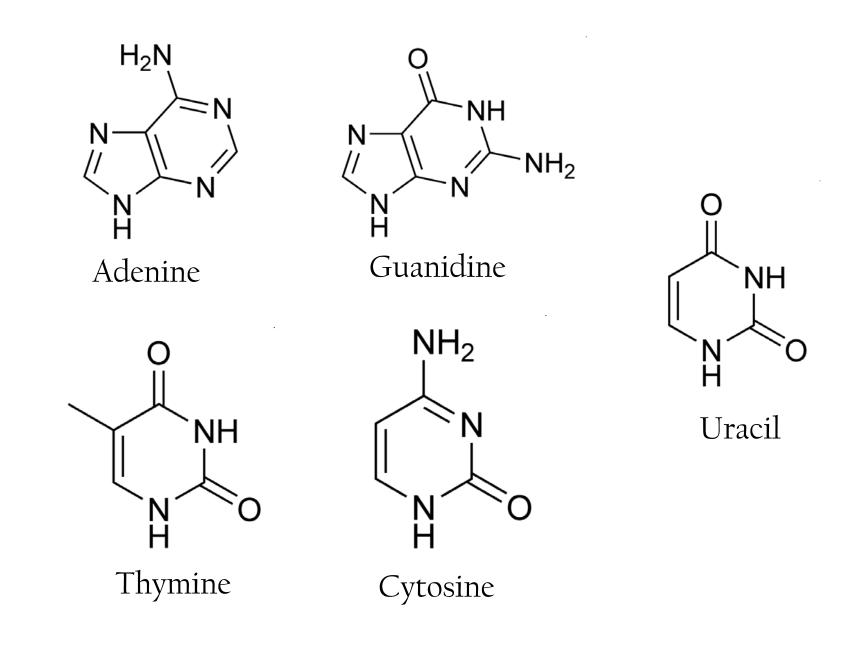
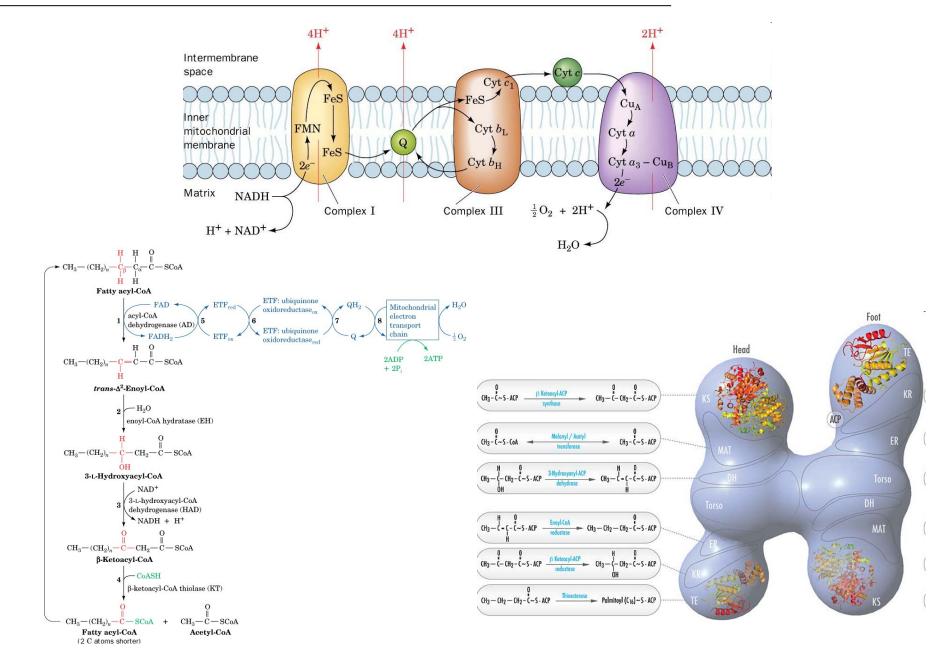
Nucleotide Metabolism

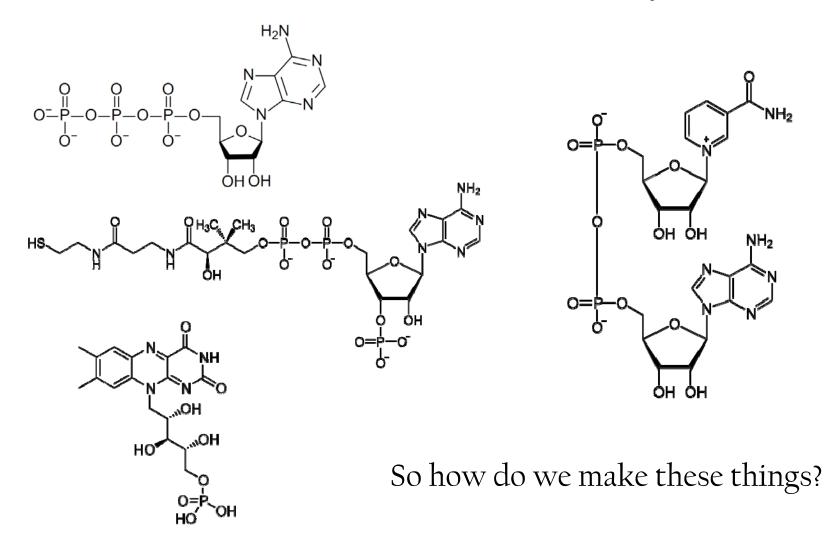


Last Week...



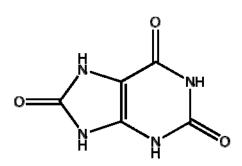
Nucleotide Metabolism

- Mostly, we think of nucleotides as being part of RNA/DNA, but now that we've studied metabolism, we know they're all over!



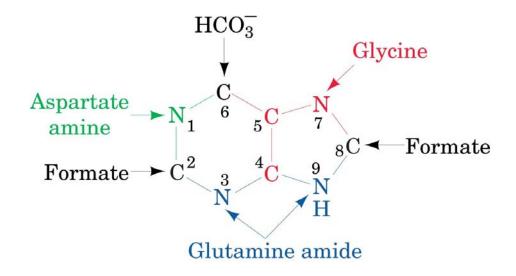
Making Purines

- Purines have a common 6,5 ring structure



Which is based on Uric Acid

- These nitrogenous rings (nitrogenous bases) are cobbled together from a number of sources:

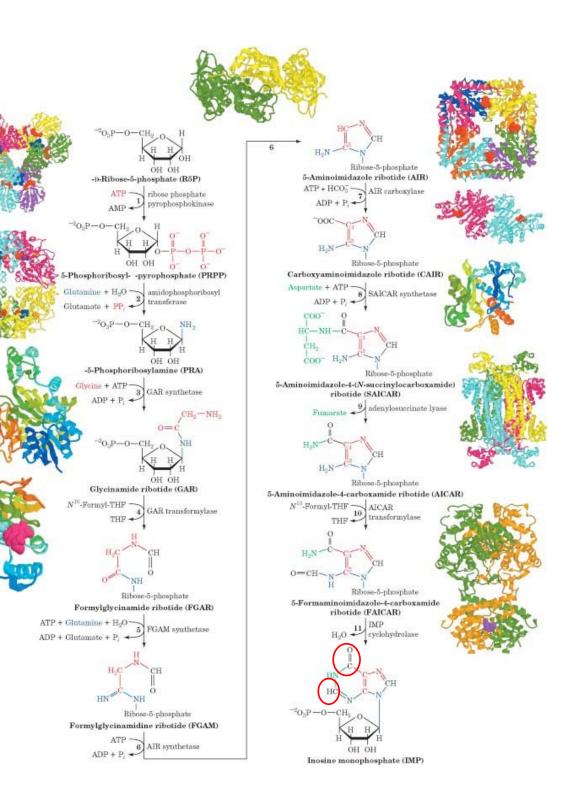


- The starting material is Ribose-5-phosphate

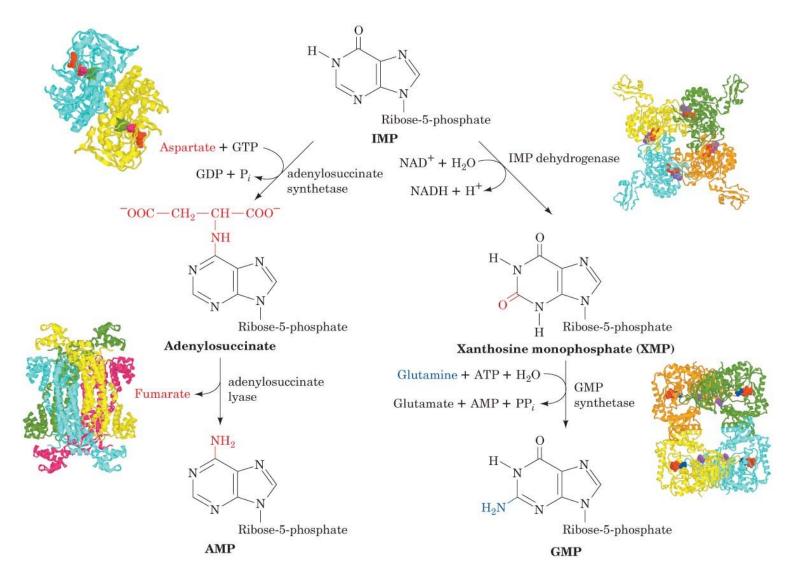
- The purine ring is actually constructed *on* the ribose

- The end product is inosine monophosphate which is similar to Adenine, except for the carbonyl at the C6 position (should be NH₂)

For Guanidine we'll need to add an NH₂ group at C2

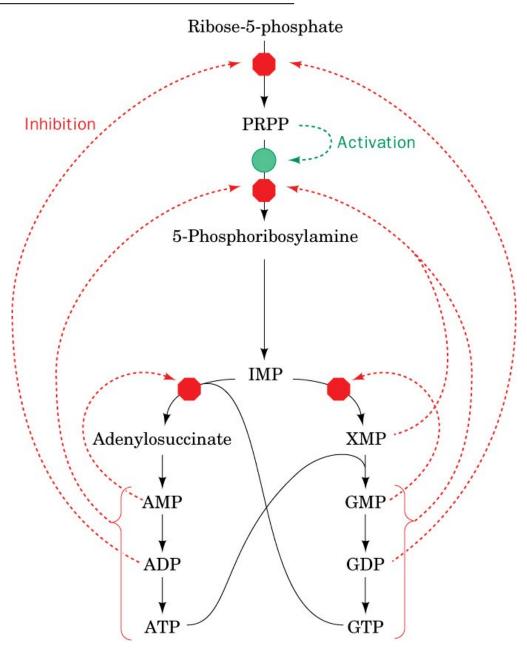


- Getting from Inosine Monophospahate to AMP and GMP



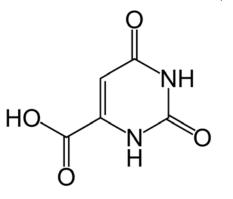
Control of Purine Synthesis

Purine synthesis is
 end-product inhibited

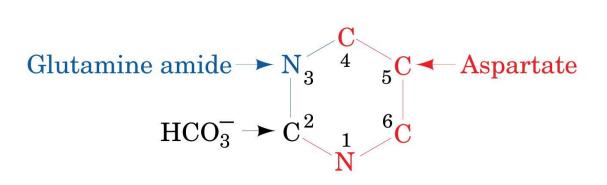


Pyridine Synthesis

- Pyridine Rings have this characteristic 6 membered ring structure typified by orotic acid (minus the COOH):

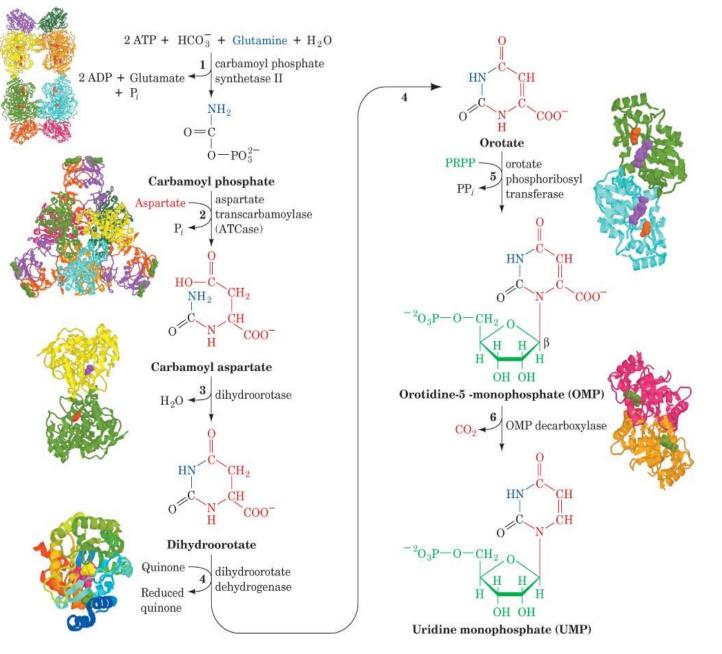


- Most of the ring comes from and Aspartate residue



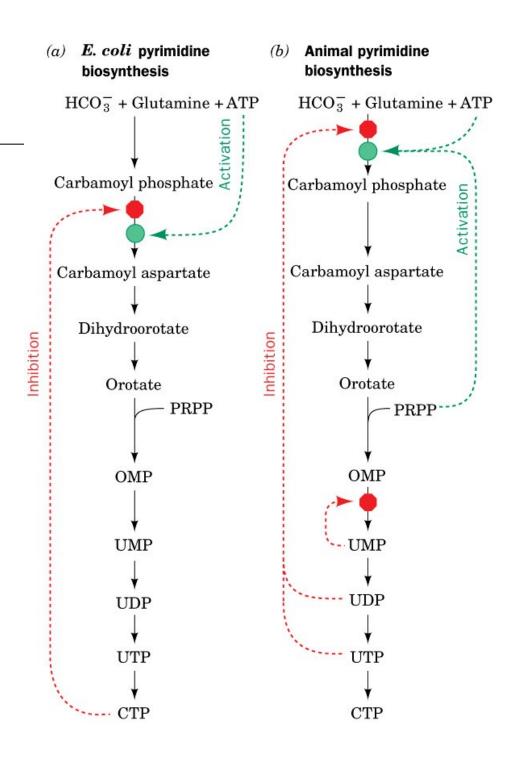
- Unlike with purines, most of the synthesis is not carried out on a ribose sugar

- PRPP (step 5) is an early intermediate in purine synthesis



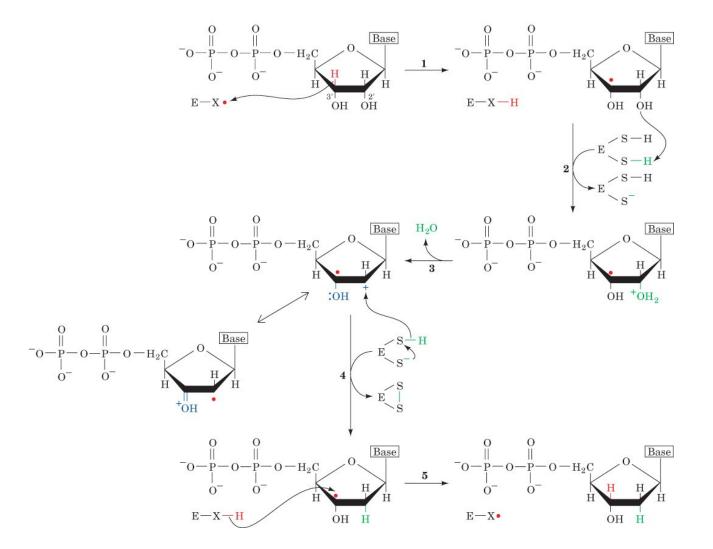
Control of Pyridine Synthesis

- Again, mostly end product inhibited

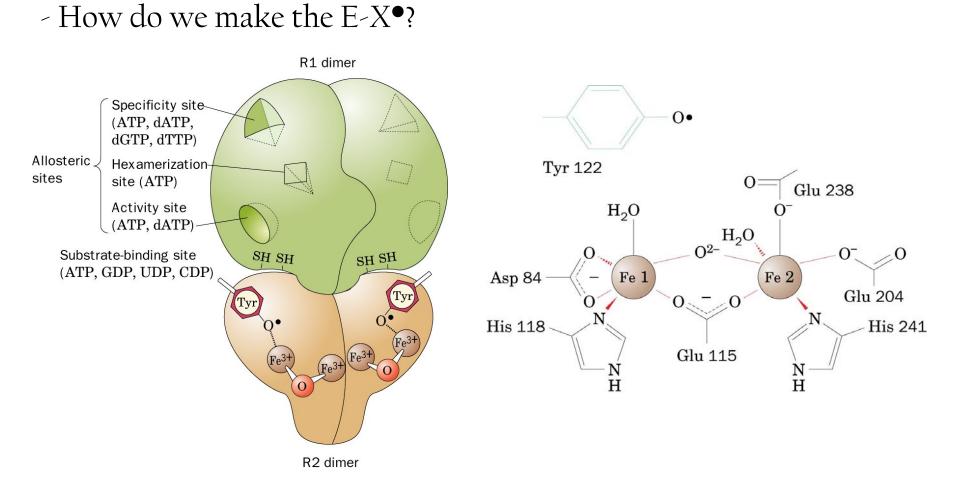


Making Deoxyribonucleotides

- Deoxynucleotides are made by Ribonucleotide Reductases which operate via a weird free radical mechanism



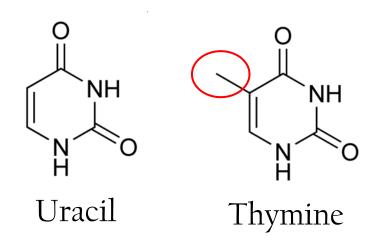
Ribonucleotide Reductases



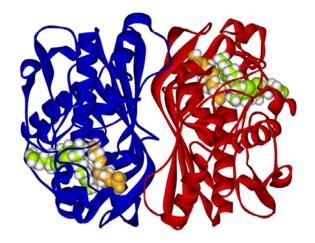
- Complete extraction of an electron from Tyr122 by Fe3⁺
- Activated by oxygen

Thymidilate Synthase: Methylation is Hard to Do

- The difference between Thymine and Uracil is a Methyl Group



- The Enzyme that makes thymine is Thyamidilate Synthase



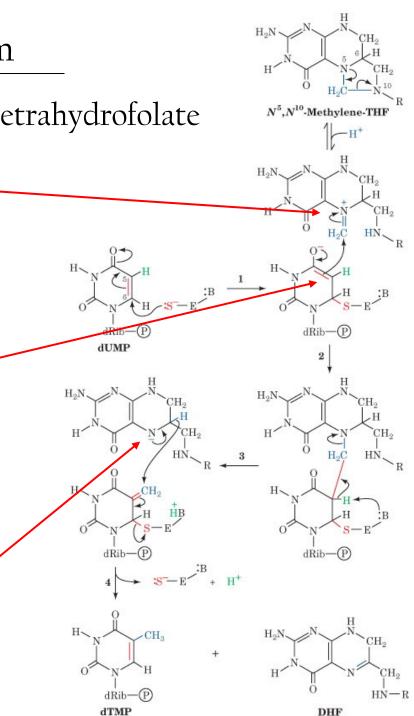
Thymidilate Synthase Mechanism

- This enzyme uses N⁵,N¹⁰-Methylenetetrahydrofolate

Step 0: Generate iminium cation by breaking 5 membered ring, attack C6 of dUMP. Moves double bond to C5/C4
Step 1: Schiff-base-like generation of a Carbocation, electrophilic attack by dUMP'

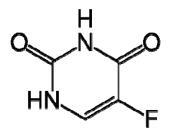
Step 2: Extraction of proton,
C=C at C5 of dUMP'

- Step 3: NADH-like hydride transfer to CH₂ on C5 of dUMP', electrons transferred back to cysteine on enzyme

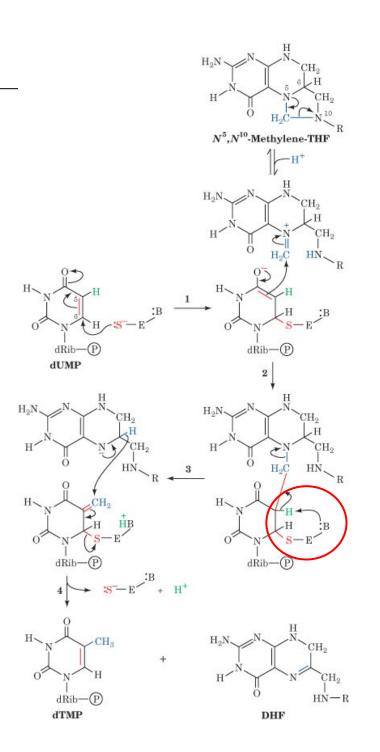


Anticancer Drug

- Fluorouracil is an anticancer drug that has been in use for over 40 years



- It is an <mark>irreversible inhibitor</mark> of Thyamine Synthase
- Interferes with step 3 because F cannot be extracted by the base.

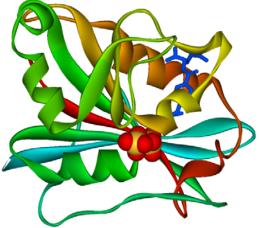


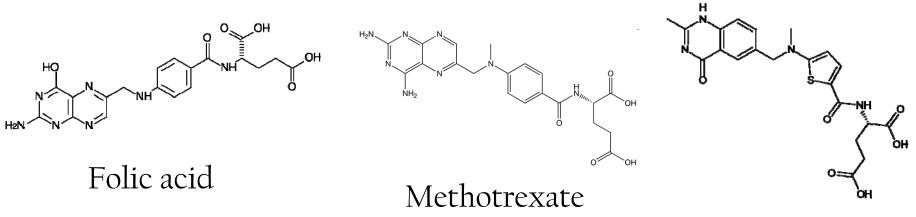
More Anticancer Drugs

- In order to keep generating dTMP, cells need to regenerate N⁵,N¹⁰-Methylenetetrahydrofolate.

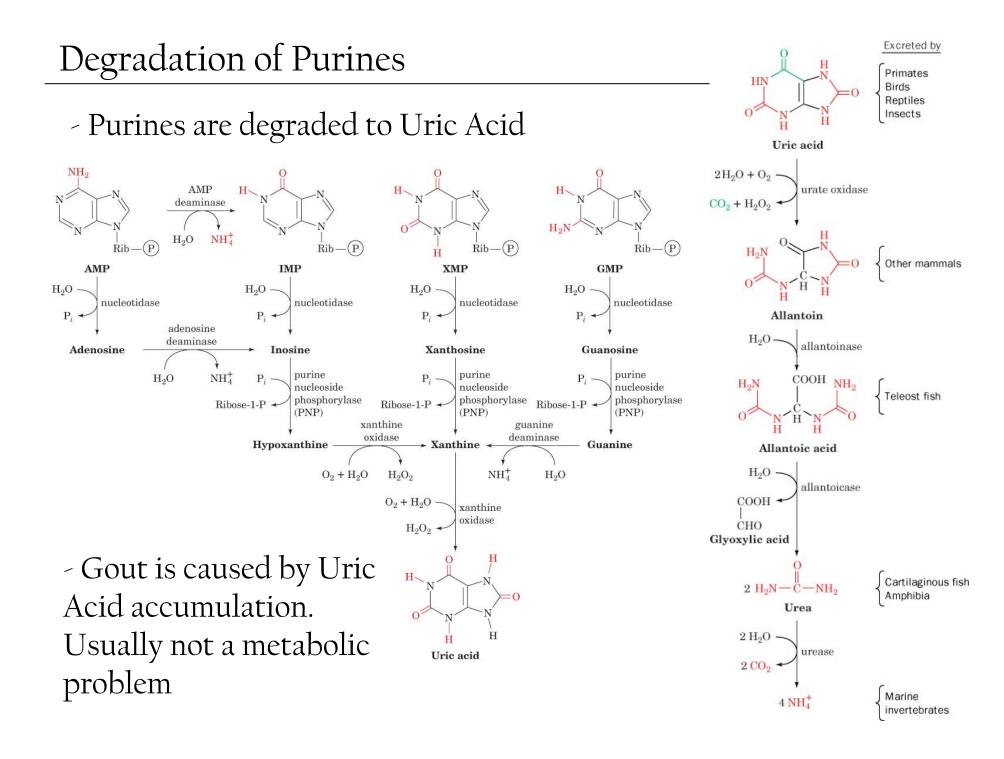
- This is normally done by dihydrofolate reductase (DHFR)

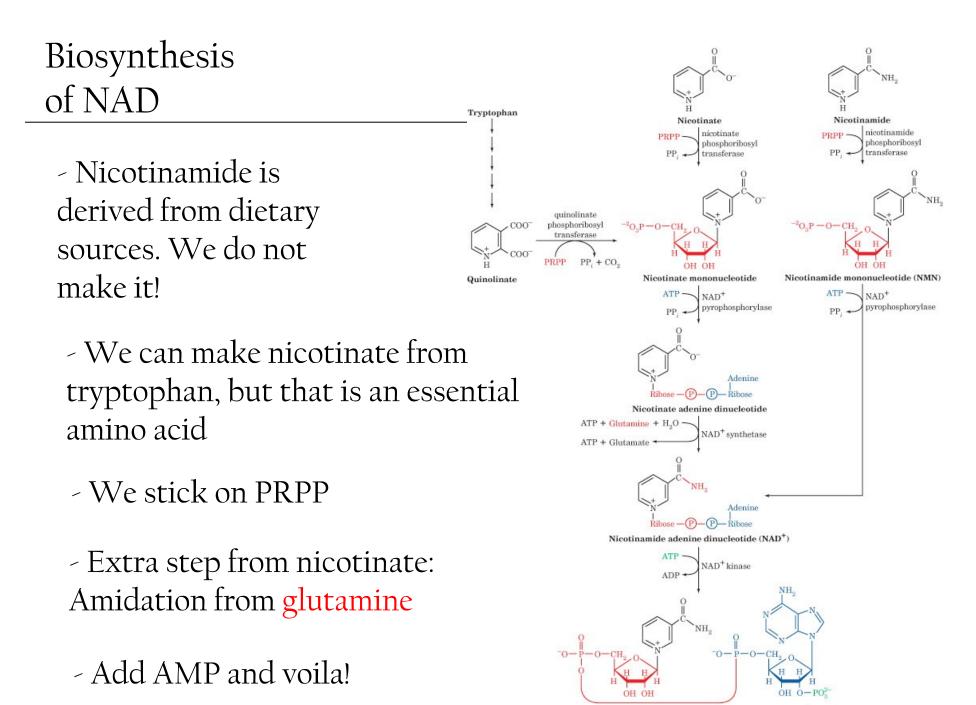
- One anti-cancer strategy is therefore to inhibit DHFR using dihydrofolate mimic inhibitors











Nicotinamide adenine dinucleotide phosphate (NADP⁺)

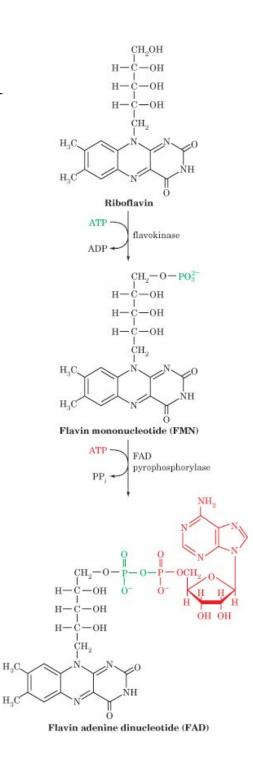


- Start with Riboflavin (vitamin B2) from dietary sources. We do not make it!

 Acticate the terminal ribose –OH group by phosphorylation (flavokinase)

 Add adenosine monophosphate (FAD pyrophosphorylase)

- Again, we're getting the oxidized form of the molecule.



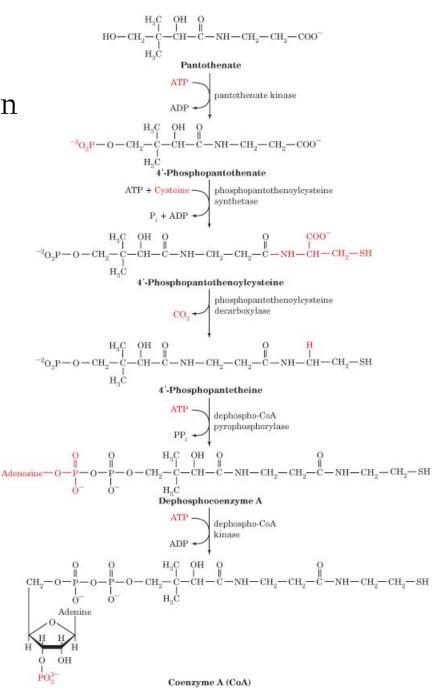
Making CoA

Start with Pantothenate (vitamin B5) which we do not make for ourselves.

- The reactive sulfur comes from cysteine!

 Removal of CO₂ by phosphopan tothenoylcysteine decarboxylase, uses a flavin cofactor

- Tack on AMP and you're done!



Linearized Michaelis-Menten Kinetics

- But how do we uncover these mechanisms?

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